A Route to the Polyacetylenic C₁₈-Acids containing 1-En-4-yne Unsaturation

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Summary A modified Wittig procedure makes possible the synthesis of 1-en-4-yne systems; ethynyl group protection with the trimethylsilyl group facilitates iodoacetylene formation for subsequent coupling reactions.

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LABELLED [1,9-¹⁴C, 9-¹⁴C, and 10-³H] methyl crepenynate (I) has been prepared¹ by the route described recently.² Attempts to use an analogous sequence of reactions for the synthesis of the more acetylenic C_{18} esters (*e.g.* II), postulated to be intermediates in the biogenetic conversion of oleate into some polyacetylenes,³ failed. Two such esters (II and III) have now been prepared in reasonable overall yields (5%) (*cf.* Bohlmann⁴) by the reaction sequence given in the Scheme. This makes possible labelling in a variety of positions, an important requirement for studies of the biogenesis of acetylenic compounds in fungi.

$$Me[CH_{2}]_{4} \cdot C \equiv C \cdot CH_{2} \cdot CH \xrightarrow{c} GH \cdot [CH_{2}]_{7} \cdot CO_{2}Me$$
(I)
$$RCH \xrightarrow{t} CH \cdot [C \equiv C]_{2} \cdot CH_{2} \cdot CH \xrightarrow{c} CH \cdot [CH_{2}]_{7} \cdot CO_{2}Me$$
(II)
$$R = Me$$
(III)
$$R = HOCH_{2}$$

$$\begin{split} \mathrm{HC} {\equiv} \mathrm{C} {\cdot} \mathrm{CH}_2 {\cdot} \mathrm{CH}_2 \mathrm{OH} \xrightarrow{\mathrm{ref.}}_{5} \\ \mathrm{Me}_3 \mathrm{Si} {\cdot} \mathrm{C} {\equiv} \mathrm{C} {\cdot} \mathrm{CH}_2 {\cdot} \mathrm{CH}_2 \mathrm{OH} \xrightarrow{\mathrm{(PhO)}_3 \mathrm{PMeI}^+}_{53\%} \\ \mathrm{Me}_3 \mathrm{Si} {\cdot} \mathrm{C} {\equiv} \mathrm{C} {\cdot} \mathrm{CH}_2 {\cdot} \mathrm{CH}_2 \mathrm{I} \xrightarrow{\mathrm{Ph}_8 \mathrm{P}/\mathrm{EtOH}}_{\mathrm{reflux}} \\ & 65\% \\ \mathrm{Me}_3 \mathrm{Si} {\cdot} \mathrm{C} {\equiv} \mathrm{C} {\cdot} \mathrm{CH}_2 {\cdot} \mathrm{CH}_2 \mathrm{PPh}_3 \mathrm{I}^- \xrightarrow{\mathrm{NaH/THF}/\mathrm{Me}_2 \mathrm{SO}}_{\mathrm{OHC} {\cdot} \mathrm{[CH}_2]_7 {\cdot} \mathrm{CO}_2 \mathrm{Me}; 0^{\circ}} \\ \mathrm{Me}_3 \mathrm{Si} \mathrm{C} {\equiv} \mathrm{C} {\cdot} \mathrm{CH}_2 {\cdot} \mathrm{CH} \xrightarrow{\mathrm{c}}_{\mathrm{CH}^+} \mathrm{CH} {\cdot} \mathrm{[CH}_2]_7 {\cdot} \mathrm{CO}_2 \mathrm{Me} \xrightarrow{\mathrm{1. AgNO}_3}_{\mathrm{2. I}_2/\mathrm{CH}_2 \mathrm{Cl}_2} \\ & 40 {-} {-} 50\% \text{ of } 95\% \text{ cis} \\ \mathrm{IC} {\equiv} \mathrm{C} {\cdot} \mathrm{CH}_2 {\cdot} \mathrm{CH} \xrightarrow{\mathrm{c}}_{\mathrm{CH}^+} \mathrm{[CH}_2]_7 {\cdot} \mathrm{CO}_2 \mathrm{Me} \xrightarrow{\mathrm{R} {\cdot} \mathrm{CH} \xrightarrow{\mathrm{t}}_{\mathrm{Ch}^+} \mathrm{CH} {\cdot} \mathrm{C} = \mathrm{CH}}_{\mathrm{Chodkiewicz}} \\ & \mathrm{(V)} \ 65\% \\ & \mathrm{(II)} \ 46\% \ (\mathrm{R} = \mathrm{Me}) \\ & \mathrm{(III)} \ 38\% \ (\mathrm{R} = \mathrm{HOCH}_2) \\ & \mathrm{Scheme} \end{split}$$

The Wittig reaction failed under normal conditions; it succeeded only when the salt (IV), the aldehyde, and sodium hydride (100%) were stirred together in tetrahydrofuran-dimethyl sulphoxide (3:2) under nitrogen at 0° for 15-50 hr. (similar reaction conditions have been previously employed to prepare $\beta\gamma$ -unsaturated acids⁶). The subsequent cleavage of the silicon-carbon bond was achieved with silver nitrate7 (cuprous chloride in aqueous ammonia was also effective) and the resulting silver acetylide was converted into the iodoacetylene (V) which was used in the coupling reaction. The trimethylsilyl-protected acetylenic phosphonium salt (IV) is likely to be of quite general utility for the synthesis of compounds containing 1-en-4-yne unsaturation.

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¹G. C. Barley, Sir Ewart R. H. Jones, V. Thaller, and R. A. Vere Hodge, unpublished work.

² R. W. Bradshaw, A. C. Day, Sir Ewart R. H. Jones, C. B. Page, and V. Thaller, *Chem. Comm.*, 1967, 1055.
³ J. D. Bu'Lock, "Comparative Phyto-chemistry," ed. T. Swain, Academic Press, London, 1966, p. 79.
⁴ F. Bohlmann, H. C. Hummel, and J. Laser, *Chem. Ber.*, 1968, 101, 3565; F. Bohlmann, R. Jente, W. Lucas, J. Laser, and H. Schulz, ibid., 1967, 100, 3194.

⁵C. Eaborn, A. R. Thompson, and D. R. M. Walton, *J. Chem. Soc.* (C), 1967, 1364; M. F. Shostakovskii, A. S. Atavin, and N. V. Egorov, *J. Gen. Chem.* (U.S.S.R.), 1965, **35**, 813. ⁶ H. S. Corey, jun., J. R. D. McCormick, and W. E. Swensen, *J. Amer. Chem. Soc.*, 1964, **86**, 1884.

7 H. M. Schmidt and J. F. Arens, Rec. Trav. chim., 1967, 86, 1138.